

**CDM-EB111-A06**

## Large-scale Methodology

---

AM0027: Substitution of CO<sub>2</sub> from fossil or mineral origin by CO<sub>2</sub> from **renewable biogenic residual** sources in the production of inorganic compounds

Version 03.0

Sectoral scope(s): 05



**United Nations**  
Framework Convention on  
Climate Change

TABLE OF CONTENTS	Page
1. INTRODUCTION .....	4
2. SCOPE, APPLICABILITY, AND ENTRY INTO FORCE .....	4
2.1. Scope.....	4
2.2. Applicability .....	4
2.3. Entry into force.....	5
2.4. Applicability of sectoral scopes .....	5
3. NORMATIVE REFERENCES .....	6
3.1. Selected approach from paragraph 48 of the CDM modalities and procedures.....	6
4. DEFINITIONS .....	6
5. BASELINE METHODOLOGY .....	6
5.1. Identification of the baseline scenario and demonstration of additionality....	6
5.1.1. <del>Step I: Identify alternatives to the project activity</del> Identification of alternative scenarios .....	7
5.1.1.1. <del>How would CO<sub>2</sub> be obtained in the absence of the CDM project activity?</del> .....	7
5.1.2. <del>Step II: Assess the project additionality and select plausible baseline candidates</del> .....	8
5.1.3. <del>Step III: Determine the most likely alternative scenario (baseline scenario)</del> .....	8
5.2. <del>Additionality</del> .....	9
5.3. Project boundary.....	9
5.4. Final use and emission reductions.....	12
5.5. <del>Baseline emissions (non-renewable CO<sub>2</sub> activity)</del> Emission reductions.....	14
5.6. <del>Project Emissions (Renewable CO<sub>2</sub> activity)</del> .....	15
5.7. <del>Emission Reduction</del> .....	16
5.8. <del>Leakage</del> .....	17
6. MONITORING METHODOLOGY .....	17
6.1. <del>Source</del> .....	17
6.2. <del>Applicability</del> .....	17

---

6.3.	Parameters not monitored .....	18
6.4.	Project emissions parameters .....	18
6.5.	Parameters monitored .....	19
6.6.	Baseline emission parameters .....	20
6.7.	Leakage .....	22

# 1. Introduction

- The following table describes the key elements of the methodology.

**Table 1. Methodology key elements**

<b>Typical projects</b>	Industrial processes where biogenic residual CO <sub>2</sub> is used as input in the production of inorganic compounds substituting CO <sub>2</sub> from fossil or mineral sources.
<b>Type of GHG emissions mitigation action</b>	Feedstock switch. Reduction of GHG emissions by using a biogenic residual source of CO <sub>2</sub> displacing fossil/mineral sources for the production of inorganic compounds.

## 2. Scope, applicability, and entry into force

### 2.1. Scope

- This methodology applies to project activities that substitute CO<sub>2</sub> obtained from fossil or mineral sources with biogenic residual CO<sub>2</sub> as input for the production of inorganic compounds.

### 2.2. Applicability

- This methodology is applicable generally to industrial production/manufacturing processes of inorganic compounds where CO<sub>2</sub> from fossil or mineral sources of CO<sub>2</sub> are presently used as an input and where renewable sources of CO<sub>2</sub> are available as a substitute input in the project activity case is substituted with biogenic residual CO<sub>2</sub>.
- The methodology is applicable under the following conditions:
  - The residual CO<sub>2</sub> from the processing of biomass was already produced but was not used before the project activity, so that no diversion of CO<sub>2</sub> from other applications is due to the project activity;
  - The processing of biomass undergoes no substantial changes in the process with the project activity;
  - CO<sub>2</sub> from fossil or mineral sources, used for the production of inorganic compounds in the baseline is from a production process whose only useful output is CO<sub>2</sub>. The CO<sub>2</sub> production process from fossil source does not produce any energy by-product;
  - CO<sub>2</sub> from fossil or mineral sources that is used for the production of inorganic compounds prior to the project activity will not be emitted to the atmosphere in the project activity;
  - There are no substantial changes (e.g. product change) in the production process of inorganic compounds as a result of the project activity;
  - Production levels of the plant (tons of inorganic compound produced per year) may in general not increase with the project activity over historic maxima;

- ~~(g) No additional significant energy quantities are required to prepare the renewable CO<sub>2</sub> from biomass processing for use in the production of inorganic compounds (related CO<sub>2</sub> emissions are below 1% of total emission reduction);~~
- ~~(h) All Carbon in the produced inorganic compounds stems from the CO<sub>2</sub> supplied during the production process.~~
- (a) Prior to the implementation of the project activity, the biogenic residual CO<sub>2</sub> was produced, but not used for any purpose;
- (b) The process generating the biogenic residual CO<sub>2</sub> undergoes no changes in capacity as a result of the project activity;
- (c) Prior to the implementation of the project activity, the fossil CO<sub>2</sub> used for the production of inorganic compounds was sourced from a process which does not involve energy production and will not continue under the project scenario;
- (d) The production process of inorganic compounds does not undergo changes, such as product, energy requirement or capacity change, as a result of the implementation of the project activity;
- (e) The biogenic residual CO<sub>2</sub> used by the project activity does not require preparation/purification measures. Any associated project emissions shall remain below 1per cent of the total emission reductions;
- (f) All the CO<sub>2</sub> used in the production of the inorganic compounds under the project activity is obtained from the biogenic residual source;
- (g) The molecules of inorganic compounds dissociate and release CO<sub>2</sub> into the atmosphere at the final use, within the crediting period of the project activity;
- (h) The baseline scenario, as determined through the procedure in section 5.1 below, is that the inorganic compounds would continue to be produced using a fossil or mineral source of CO<sub>2</sub> and that the biogenic residual CO<sub>2</sub> would continue to be produced but not used for any purpose.

~~5. This baseline methodology shall be used in conjunction with the approved monitoring methodology AM0027 (Substitution of CO<sub>2</sub> from fossil or mineral origin by CO<sub>2</sub> from renewable sources in the production of inorganic compounds).~~

## 2.3. Entry into force

6. The date of entry into force of the methodology is the date of the publication of the EB 111 meeting report on 9 September 2021.

## 2.4. Applicability of sectoral scopes

7. For validation and verification of CDM projects and programmes of activities by a designated operational entity (DOE) using this methodology, application of sectoral scope 05 is mandatory.

### 3. Normative references

8. This methodology is based on the part called "renewable CO<sub>2</sub> activity" of the project activity "Raudi Chemical Salts", proposed by Raudi Indústria e Comércio Ltda., with participation of Coopcana – Cooperativa Agrícola Regional de Produtores de Cana Ltda, whose baseline study and project design document were prepared by Ecoinvest, Brazil.
9. This methodology is based on elements from the proposed new methodology NM0115: "CO<sub>2</sub>, electricity and steam from renewable sources in the production of inorganic compounds".
10. For more information regarding the proposal and its consideration by the Executive Board (hereinafter referred as the Board) of the clean development mechanism (CDM) please refer to case NM0115: "CO<sub>2</sub>, electricity and steam from renewable sources in the production of inorganic compounds" on <<http://cdm.unfccc.int/methodologies/PAmethodologies/approved.html>>.
11. This methodology also refers to the latest version of the "Tool for the demonstration and assessment of additionality".<sup>†</sup> This methodology also refers to the latest approved versions of "TOOL02: Combined tool to identify the baseline scenario and demonstrate additionality"

#### 3.1. Selected approach from paragraph 48 of the CDM modalities and procedures

12. "Emissions from a technology that represents an economically attractive course of action, taking into account barriers to investment".

### 4. Definitions

13. The definitions contained in the Glossary of CDM terms shall apply.
14. For the purpose of this methodology, the following definitions apply:
  - (a) **Biogenic residual CO<sub>2</sub>** – stream of CO<sub>2</sub> that is generated through a biological process (e.g. fermentation of biomass) and emitted to the atmosphere in the absence of the project activity;
  - (b) **CO<sub>2</sub> from fossil/mineral sources** – stream of CO<sub>2</sub> that is obtained from fossil/mineral sources.

### 5. Baseline methodology

#### 5.1. Identification of the baseline scenario and demonstration of additionality

15. The methodology determines the baseline scenario through the following steps:
  - (a) Step I: Identify alternatives to the project activity;
  - (b) Step II: Assess project additionality;

<sup>†</sup> Please refer to: <<http://cdm.unfccc.int/methodologies/PAmethodologies/approved.html>>.

(c) ~~Step III: Determine the most likely scenario (baseline scenario) from among the alternatives identified.~~

16. The selection of the baseline scenario and demonstration of additionality shall be conducted in accordance with "TOOL02: Combined tool to identify the baseline scenario and demonstrate additionality".

#### 5.1.1. ~~Step I: Identify alternatives to the project activity~~ Identification of alternative scenarios

17. Project participants shall identify realistic and credible alternatives(s) to each one of the project components in the country where the inorganic compound is produced and consumed. Alternatives should be separately determined regarding scenarios, including:

(a) ~~How would CO<sub>2</sub> be obtained in the absence of the CDM project activity?~~

(b) ~~What would happen to the primary source of renewable energy in the absence of the project activity?~~

(c) ~~What would happen to the baseline and project sources of CO<sub>2</sub> in the absence of the project activity?~~

(a) All possible sources of CO<sub>2</sub> for the production of inorganic compounds;

(b) The fate of the process generating the biogenic residual CO<sub>2</sub> used under the project activity;

(c) The fate of the process that generated the fossil/mineral CO<sub>2</sub> prior to the implementation of the project activity.

##### 5.1.1.1. ~~How would CO<sub>2</sub> be obtained in the absence of the CDM project activity?~~

18. ~~The production of some inorganic compounds requires CO<sub>2</sub> as raw material. The gas reacts with other raw material inside a chemical reactor producing the final product. CO<sub>2</sub> can be obtained from fossil, mineral or renewable sources.~~

19. ~~Alternatives to the renewable CO<sub>2</sub> activity (the project activity) must be identified through the consultation of technical associations and official (country) information, supported by technical literature or market researches, as appropriate. The alternatives are to be identified inside the country where the project is developed.~~

20. ~~There is no single standardized source for this type of information, hence, the sources of information must be determined on a project specific basis and evaluated during the validation phase, by the DOE. The sources selected must be well recognized and widespread accepted as a reliable source in the country where the project is developed.~~

21. ~~For CO<sub>2</sub>, the realistic and credible alternative(s) may include, *inter alia*: Amongst the realistic and credible alternatives, the following scenarios (and any plausible combination) shall be considered:~~

(a) C1: The proposed project activity ~~(use of renewable source of CO<sub>2</sub>)~~ not undertaken as a CDM project activity;

(b) C2: The proposed project activity, implemented at a later point in time and not undertaken as a CDM project activity;

- (c) C3: The proposed project activity, using the same type of renewable biogenic residual CO<sub>2</sub> but with a lower/higher CO<sub>2</sub> consumption (e.g. CO<sub>2</sub> use efficiency that is common practice in the relevant industry sector);
- (d) C4: The use of CO<sub>2</sub> from a particular existing or new plant, on-site or off-site, using other renewable or biogenic residual CO<sub>2</sub> sources, such as other biomass sources;
- (e) C5: The use of CO<sub>2</sub> from a particular existing or new plant, on-site or off-site, using non-renewable sources of CO<sub>2</sub>, such as CO<sub>2</sub> derived from thermochemical processing of fossil hydrocarbons<sup>2</sup>, CO<sub>2</sub> derived from mineral products<sup>3</sup>, etc. If not used as input for the production of inorganic compounds, the CO<sub>2</sub> would not be produced, and would not be emitted in the atmosphere fossil/mineral sources without energy outputs;
- (f) C6: The use of CO<sub>2</sub> in a particular existing or new plant, on-site or off-site, using non-renewable residual CO<sub>2</sub> sources, such as residual CO<sub>2</sub> from other industrial processes that uses fossil or mineral as raw materials, as in the cement industry. If not used as input for the production of inorganic compounds, the CO<sub>2</sub> would accrue anyway and would be emitted in the atmosphere.

22. Note that the alternatives proposed in this Section are only indicative. Project proponents may propose other possible alternatives and/or eliminate some of the proposed above, based on documented evidence.

#### 5.1.2. Step II: Assess the project additionality and select plausible baseline candidates

23. Project participants, after identifying the alternatives to and building realistic and credible scenarios shall apply the latest approved version of the "Tool for the demonstration and assessment of additionality" for two purposes: (i) to assess project scenario additionality, showing that it is not part of the baseline; and (ii) to identify which one of the alternatives should be excluded from further consideration for baseline determination (e.g. alternatives where barriers are prohibitive or which are clearly economically unattractive). The Tool shall be applied without any modification for the set of alternative scenarios identified.

#### 5.1.3. Step III: Determine the most likely alternative scenario (baseline scenario)

24. Where more than one credible and plausible alternative scenario remains, project participants shall, as a conservative assumption, use the alternative baseline scenario that results in the lowest baseline emissions as the most likely baseline scenario.

25. This methodology is only applicable to a project activity if it can be demonstrated through steps I to III above that alternative C5 is the most likely baseline scenario.

<sup>2</sup> Fossil origin, by thermochemical processing of synthesis gas (methane, for example) or of other hydrocarbons derived from petrochemical industry. CO<sub>2</sub> that is purchased from suppliers of industrial gases normally stems from thermochemical processing.

<sup>3</sup> Mineral origin, whether from the calcination of calcium carbonate (CaCO<sub>3</sub>), as in the Solvay process, renowned as the most usual fabrication process of sodium bicarbonate worldwide, or directly obtained from mineral ore that contains the inorganic compound.



## 5.2. **Additionality**

26. The additionality of the project activity shall be demonstrated and assessed using the latest version of the “**Tool for the demonstration and assessment of additionality**” agreed by the CDM Executive Board, available at the UNFCCC CDM web site<sup>4</sup>.

## 5.3. **Project boundary**

27. For the purpose of determining GHG emissions of the **baseline**, project participants shall include the following emissions sources:

- (a) GHG emissions from on-site production of inorganic compounds (if any); and
- (b) GHG emissions from final use of inorganic compounds (if any).

28. For the purpose of determining the GHG emissions of the **project activity**, project participants shall include the following emission sources:

- (a) Sequestration of CO<sub>2</sub> from the atmosphere (if any);
- (b) GHG emissions from on-site processing of biomass that generates the residual CO<sub>2</sub> (if any); and
- (c) GHG emissions from on-site production of inorganic compounds (if any); and
- (d) GHG emissions from final use of inorganic compounds (if any).

29. Figure 1 shows the baseline boundary. Figure 42 provides an overview on the project activity boundary, Figure 2 shows the baseline boundary.

---

<sup>4</sup> Please refer to: < <http://cdm.unfccc.int/methodologies/PAmethodologies/approved.html> >.

Figure 21. Baseline boundary

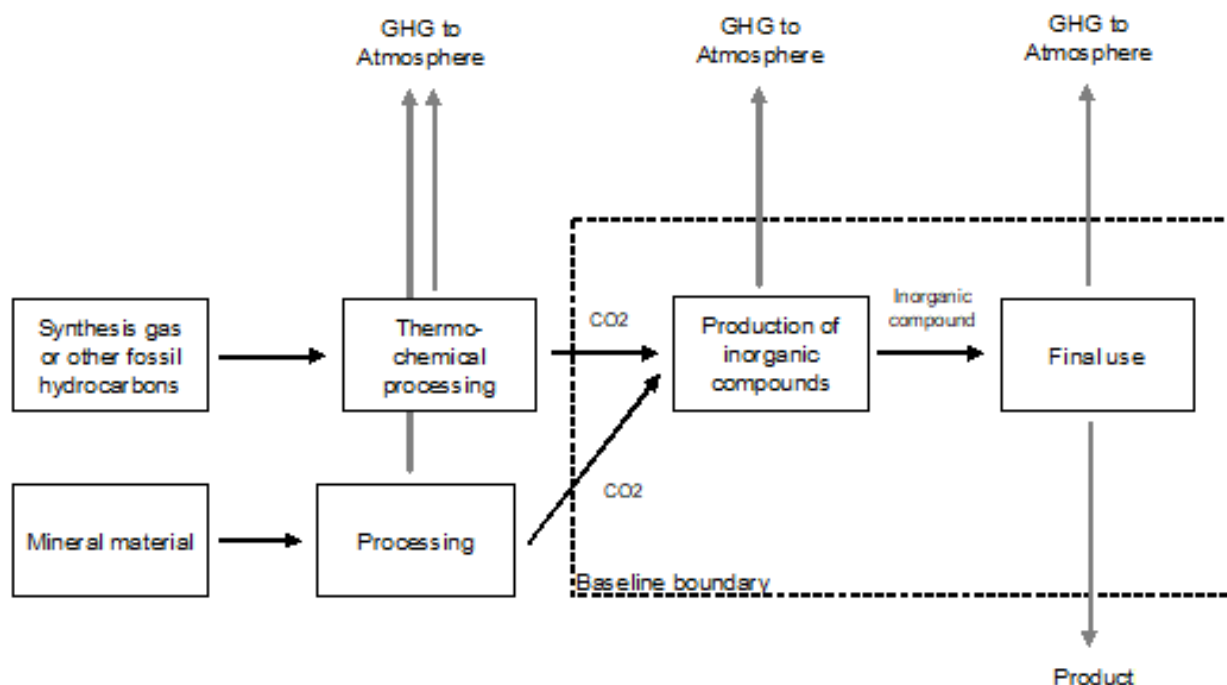
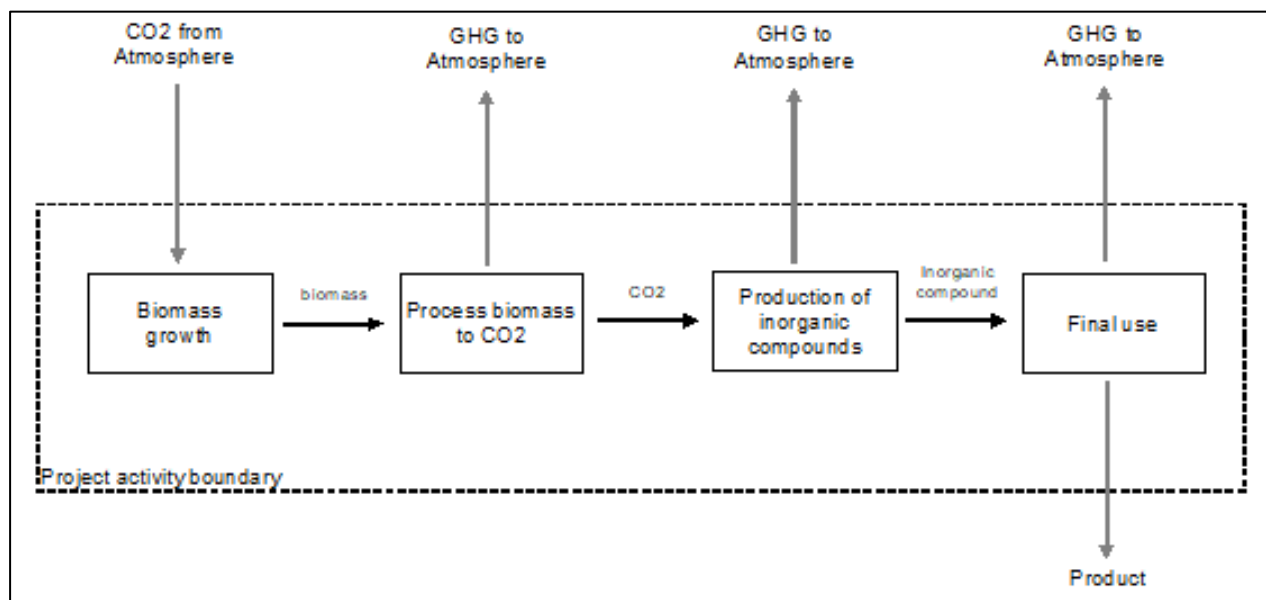


Figure 12. Project activity boundary



30. Table 2 illustrates in detail which emissions sources are included and which are excluded from the project boundary for determination of both baseline and project emissions.

**Table 2. Emission sources included in or excluded from the project boundary**

Source		Gas	Included	Justification/Explanation
Baseline	Processing of fossil or mineral hydrocarbons Production of CO <sub>2</sub> from fossil/mineral sources	CO <sub>2</sub>	Yes	Main source of emissions
		CH <sub>4</sub>	No	
		N <sub>2</sub> O	No	
	Production of inorganic compounds Biogenic process producing CO <sub>2</sub>	CO <sub>2</sub>	Yes (but cancelled out)	Due to applicability criteria, the process of the production of inorganic compounds is not changed with the implementation of the project activity, and potential baseline and project emissions are the same Main source of emissions
		CH <sub>4</sub>	No	Excluded for simplification
		N <sub>2</sub> O	No	Excluded for simplification
	Final use	CO <sub>2</sub>	Yes	CO <sub>2</sub> is either emitted to the atmosphere or stored
		CH <sub>4</sub>	No	Excluded for simplification
		N <sub>2</sub> O	No	Excluded for simplification

Source		Gas	Included	Justification/Explanation
Project activity	Uptake of CO <sub>2</sub> by biomass growth	CO <sub>2</sub>	Yes (but included elsewhere)	CO <sub>2</sub> uptake in biomass growth is covered by the fact that renewal CO <sub>2</sub> is treated climate-neutral
		CH <sub>4</sub>	No	
		N <sub>2</sub> O	No	
	Processing of biomass, producing residual CO <sub>2</sub> Biogenic process producing CO <sub>2</sub>	CO <sub>2</sub>	Yes No (but cancelled out)	May be an important emission source. However, due to applicability criteria, the biomass processing is not changed with the implementation of the project activity, and baseline and project emissions are the same. In accordance with the applicability conditions, project emissions from the biogenic process are considered to remain the same as in the baseline
		CH <sub>4</sub>	No	Excluded for simplification
		N <sub>2</sub> O	No	Excluded for simplification
	Production of inorganic compounds Production of CO <sub>2</sub> from fossil/mineral sources	CO <sub>2</sub>	Yes (but cancelled out) No	Due to applicability criteria, the process of the production of inorganic compounds is not changed with the implementation of the project activity, and potential baseline and project emissions are the same. All the carbon content of the inorganic compounds produced under the project activity is sourced from the biogenic residual CO <sub>2</sub> supplied by the project activity
		CH <sub>4</sub>	No	Excluded for simplification
		N <sub>2</sub> O	No	Excluded for simplification
	Final use	CO <sub>2</sub>	Yes	CO <sub>2</sub> is either emitted to the atmosphere or stored
		CH <sub>4</sub>	No	Excluded for simplification
		N <sub>2</sub> O	No	Excluded for simplification

#### 5.4. Final use and emission reductions

31. The project activity reduces net CO<sub>2</sub> emissions to the atmosphere or leads to C sequestration by substituting CO<sub>2</sub> from fossil or mineral origin sources with biogenic residual CO<sub>2</sub> that originates from the processing of biomass as input for the production process of inorganic compounds.
32. In the final use, the inorganic compounds may either (i) thermally dissolve or (ii) not dissociate. The project activity does not affect the downstream fate of the produced inorganic compounds. Therefore, it may be assumed that the final use of the product would be the same in the baseline and project scenarios.

(a) Assuming that the inorganic compound molecules thermally dissolve in the final use. Hence, if a project activity uses renewable CO<sub>2</sub> instead of non-renewable CO<sub>2</sub> of fossil or mineral origin, emissions of non-renewable CO<sub>2</sub> during the final use of the compound are avoided.

(b) On the other hand, in the case the inorganic compound molecules do not dissociate during the final use, the result of the project activity is carbon sequestration, because CO<sub>2</sub> is continuously sequestered from the atmosphere by the production of inorganic chemical. Hence, the project activity leads to the permanent removal of CO<sub>2</sub> from the atmosphere (or "negative" emissions).

33. It is important to note, however, that the main objective of the "renewable CO<sub>2</sub> activity" is not to sequester CO<sub>2</sub> from the atmosphere. The point about sequestration is to demonstrate that, even in the case some portion of the chemical doesn't dissociate during the final use phase, the activity continues to lead to emissions reductions.

34. The renewable CO<sub>2</sub> may be obtained from the processing of biomass, e.g. from the waste CO<sub>2</sub> stream, which was previously released to the atmosphere, generated in the fermentation of sugar cane juice for the production of ethanol.

35. In Table 3, representations of both situations, with and without dissociation in the final use, are presented.

**Table 3. Emissions balance in the "renewable CO<sub>2</sub> activity"**

		Does the compound release CO <sub>2</sub> in the final use?	What happens to emissions in the project activity scenario?	What would happen to emissions in the baseline scenario?	Emissions balance, from the difference between baseline and project emissions
Project scenario with renewable CO <sub>2</sub> and baseline scenario with non-renewable CO <sub>2</sub>	<b>SITUATION 1 With dissociation</b>	Yes, compound molecule dissociates and CO <sub>2</sub> is emitted to the atmosphere in the final use	CO <sub>2</sub> emissions occur and are renewable (net emission is zero)	Fossil or mineral CO <sub>2</sub> emissions would occur (net emission would be positive)	Emissions reductions occur due to the avoidance of non-renewable CO <sub>2</sub> emissions
	<b>SITUATION 2 Without dissociation</b>	No, compound molecule does not dissociate and CO <sub>2</sub> is not emitted to the atmosphere in the final use	Atmosphere CO <sub>2</sub> is sequestered by the chemical molecule (net emission is negative)	Fossil or mineral CO <sub>2</sub> would be sequestered by the chemical molecule (net emission would be zero)	Emissions reductions occur due to the sequestration of atmosphere CO <sub>2</sub> by the chemical molecule

36. Please note that, switching from one situation to the other does not affect the net emission reductions of the project, as it may be assumed that the final use of the product in the baseline would be the same as in the project activity. It means that, if in the project activity scenario CO<sub>2</sub> is released in the final use, it would also happen in the baseline scenario

and vice-versa, and the emissions reductions would be the same, independently of the situation describing the final use.

## 5.5. Baseline emissions (non-renewable CO<sub>2</sub> activity) Emission reductions

37. If the baseline scenario C.5 is identified (see Section "Identification of baseline scenario"), i.e. the production of the inorganic compound with fossil or mineral CO<sub>2</sub> sources, then emissions of the baseline due to the final use of the inorganic compounds are calculated as below.

38. When the final use of the inorganic compound emits CO<sub>2</sub> to the atmosphere, the emissions from the final use of the inorganic compounds are  $N$  moles of CO<sub>2</sub> for each mol of inorganic compound used. Hence, the following emission factor for the "CO<sub>2</sub>-activity" results [in tCO<sub>2</sub>/t of inorganic compound]:

$$EF_{CA} = 44(N/M) \quad \text{Equation (1)}$$

Where:

44	=	The molecular weight of CO <sub>2</sub> , [g/mol]
$N$	=	The carbon content of the inorganic compound, i.e., the number of carbon atoms in the inorganic compound molecule that would thermally dissociate in the final use of one molecule of the compound. $N$ is a fixed parameter that depends on the inorganic compound involved, [non dimensional]
$M$	=	The molecular weight of the inorganic compound, a fixed parameter that depends on the inorganic compound involved. It is calculated straightforwardly by summing the atomic weights of the compound constituents, in [g]

39. This is based on the assumption that all Carbon in the produced inorganic compounds stems from the CO<sub>2</sub> supplied during the production process (applicability criterium).

40. The calculation of the baseline emissions ( $B$ ) consists of three parts, the GHG emission ( $BE$ ) and sequestration ( $BS$ ) parts during final consumption, as well as possible emissions related to the activity, e.g. from the production of inorganic compounds ( $BI$ ). It is calculated as follows. Considering that methodology is only applicable if the production process of inorganic compounds does not undergo changes of product, energy requirement or capacity, it is assumed that project and leakage emissions are not affected by the implementation of the project activity. Therefore, emission reductions are calculated based on a simplified approach as follows:

$$B = BE - BS + BI \quad \text{Equation (2)}$$

$$ER_y = \min(P_{his}, P_y) \times \min(Q_{CO2,his}, Q_{CO2,y}) \quad \text{Equation (1)}$$

Where:

$P_{his}$	=	Average amount of inorganic compound produced per year in the last three years prior to the implementation of the project activity (t)
-----------	---	--

$P_y$  = Amount of inorganic compound produced in year y (t)

$Q_{CO_2,his}$  = Average amount of fossil/mineral CO<sub>2</sub> used per tonne of inorganic compound produced in the last three years prior to the implementation of the project activity (tCO<sub>2</sub>/ t product)

$Q_{CO_2,y}$  = Amount of biogenic residual CO<sub>2</sub> used per tonne of inorganic compound produced in year y (tCO<sub>2</sub>/ t product)

41. This is based on the assumption that all Carbon in the produced inorganic compounds stems from the CO<sub>2</sub> supplied during the production process (applicability criterium).

42. The emissions of non-renewable CO<sub>2</sub> are are:

$$BE = EF_{CA} m_{\pm} (1 - k_B) \quad \text{Equation (3)}$$

Thus,

$$BE = 44(N/M) m_{\pm} (1 - k_B) \quad \text{Equation (4)}$$

43. The sequestration in the baseline is

$$BS = EF_{CA} m_{\pm} k_B \quad \text{Equation (0)}$$

Thus,

$$BS = 44(N/M) m_{\pm} k_B \quad \text{Equation (5)}$$

44. Definition of parameters and variables is provided at the end of the project emissions section below.

## 5.6. Project Emissions (Renewable CO<sub>2</sub> activity)

45. The calculation of the project emissions ( $P$ ) also consists of three parts, the GHG emission ( $PE$ ) and sequestration ( $PS$ ) parts during final consumption, as well as other possible emissions related to the activity, e.g. from the production of inorganic compounds ( $PI$ ). It is project emissions are calculated as follows:

$$P = PE - PS + PI \quad \text{Equation (6)}$$

46. The emissions of non-renewable CO<sub>2</sub> are

$$PE = EF_{CA} m_{\pm} (1 - k_P) \quad \text{Equation (7)}$$

Thus,

$$PE = 44(N/M)m_z(1 - k_p) \quad \text{Equation (8)}$$

47. The sequestration in the project activity is

$$PS = EF_{CA}m_zk_p \quad \text{Equation (9)}$$

Thus,

$$PS = 44(N/M)m_zk_p \quad \text{Equation (10)}$$

Where:

$m_1$	=	The amount of the inorganic compound that releases CO <sub>2</sub> in the final use, in [t]
$m_2$	=	The amount of the inorganic compound that does <i>not</i> release CO <sub>2</sub> in the final use, in [t]
$m = m_1 + m_2$	=	The total amount of the inorganic compound produced, in [t]
$k_b$ and $k_p$	=	Non-dimensional correction factors for renewable CO <sub>2</sub> in the baseline and project activity and are calculated as follows:

$$k_b = m_{br}/(m_{br} + m_{bnr}) \text{ and } k_p = m_{pr}/(m_{pr} + m_{pnr})$$

Where:

$m_{br}$	=	$m_{br}$ is the total amount of renewable CO <sub>2</sub> used in the baseline scenario
$m_{bnr}$	=	$m_{bnr}$ is the total amount of non-renewable CO <sub>2</sub> used in the baseline scenario
$m_{pr}$	=	$m_{pr}$ is the total amount of renewable CO <sub>2</sub> used by the project
$m_{pnr}$	=	$m_{pnr}$ is the total amount of non-renewable CO <sub>2</sub> used by the project

## 5.7. Emission Reduction

48. The applicability criteria (see above) require that neither the processing of biomass (generating waste renewable CO<sub>2</sub>) nor the production process (for inorganic compounds) undergo substantial changes with the project activity. With this, it may be assumed that potential GHG emissions from the biomass processing and the production process remain the same in the baseline (*BI*) and in the project activity (*PI*) case:

$$BI = PI \quad \text{Equation (11)}$$

49. The total emission reduction which covers direct emissions of GHG and sequestration can be written as:

$$ER = B - P = (BE - PE) + (PS - BS) + (BI - PI) \quad \text{Equation (12)}$$



$$ER = (BE - PE) + (PS - BS) \quad \text{Equation (13)}$$

$$ER = 44(N/M)(m_x + m_y)(k_p + k_B) \quad \text{Equation (14)}$$

$$ER = 44(N/M) m (k_p + k_B) \quad \text{Equation (15)}$$

## 5.8. Leakage

50. The main potential source of leakage for this project activity lies in an increase in emissions due to diversion of CO<sub>2</sub> from other users to the project as a result of the project activity.

51. This source of leakage is zero if the conditions under which the methodology is applicable are satisfied:

(a) The residual CO<sub>2</sub> from the processing of biomass was already produced but was not used before the project activity, so that no diversion of CO<sub>2</sub> from other applications is due to the project activity.

52. Project participants must provide adequate evidence of this condition during the validation of the project activity.

## 6. Monitoring Methodology

### 6.1. Source

53. This methodology is based on the part called "renewable CO<sub>2</sub> activity" of the project activity "Raudi Chemical Salts", proposed by Raudi Indústria e Comércio Ltda., with participation of Coopcana – Cooperativa Agrícola Regional de Produtores de Cana Ltda, whose baseline study and project design document were prepared by Mr. Rodrigo Marcelo Leme, from Ecoinvest, Brazil.

54. For more information regarding the proposal and its consideration by the Executive Board please refer to case NM0115: "CO<sub>2</sub>, electricity and steam from renewable sources in the production of inorganic compounds" on <<http://cdm.unfccc.int/methodologies/PAmethodologies/approved.html>>.

55. This methodology also refers to the latest version of the "Tool for the demonstration and assessment of additionality"<sup>5</sup>.

### 6.2. Applicability

56. This monitoring methodology shall be used in conjunction with the approved baseline methodology AM0027 (Substitution of CO<sub>2</sub> from fossil or mineral origin by CO<sub>2</sub> from renewable sources in the production of inorganic compounds). The same applicability conditions as in baseline AM0027 apply.

<sup>5</sup> Please refer to: <<http://cdm.unfccc.int/methodologies/PAmethodologies/approved.html>>.

57. All data collected as part of monitoring should be archived electronically and be kept at least for 2 years after the end of the last crediting period. 100% of the data should be monitored if not indicated differently in the comments in the tables below.

### 6.3. Parameters not monitored

### 6.4. Project emissions parameters

58. The following table illustrates the data to be collected or used in order to monitor emissions from the project and baseline activity.

**Data / Parameter table 1.**

Data / Parameter:	N
Data unit:	Non dimensional
Description:	Carbon content of the inorganic compound, i.e., the number of carbon atoms in the inorganic compound molecule that would thermally dissociate in the final use of one molecule of the compound
Source of data:	Technical literature, such as, chemical engineers handbooks
Measurement procedures (if any):	
Monitoring frequency:	Once, at the validation
QA/QC procedures:	Check consistency with literature
Any comment:	This is a fixed parameter that needs to be demonstrated through the chemical dissociation equation in the final use of each compound produced

**Data / Parameter table 2.**

Data / Parameter:	M
Data unit:	g/mol
Description:	Molecular weight of the inorganic compound.
Source of data:	Technical literature, such as, chemical engineers handbooks
Measurement procedures (if any):	
Monitoring frequency:	Once, at the validation
QA/QC procedures:	Check consistency with literature
Any comment:	This is a fixed parameter calculated from the summation of the atomic weights of the compound constituents

**Data / Parameter table 1.**

Data / Parameter:	$mP_{his}$
Data unit:	t
Description:	Average amount of inorganic compound produced per year in the last three years prior to the implementation of the project activity
Source of data:	On-site records

Measurement procedures (if any):	Use calibrated mass or volume meters
Monitoring frequency:	Monthly
QA/QC procedures:	To be cross-checked with sales receipts and an energy balance that is based on purchased quantities and stock changes
Any comment:	

**Data / Parameter table 2.**

<b>Data / Parameter:</b>	<b><math>Q_{CO_2, his}</math></b>
Data unit:	tCO <sub>2</sub> / t product
Description:	Average amount of fossil/mineral CO <sub>2</sub> used per tonne of inorganic compound produced in the last three years prior to the implementation of the project activity
Source of data:	Project site records
Measurement procedures (if any):	
Monitoring frequency:	The amount CO <sub>2</sub> per tonne of inorganic compound produced in the baseline shall be monitored for at least three years before the start of the project activity
QA/QC procedures:	
Any comment:	Based on the carbon content (i.e. the number of carbon atoms in the compound molecule) and molecular weight of the inorganic compound produced in the baseline

## 6.5. Parameters monitored

**Data / Parameter table 3.**

<b>Data / Parameter:</b>	<b><math>mP_y</math></b>
Data unit:	t
Description:	Total amount of chemical inorganic compound produced in year y
Source of data:	Company books, sales documents, On-site records
Measurement procedures (if any):	Use calibrated mass or volume meters
Monitoring frequency:	Monthly
QA/QC procedures:	<del>Any direct measurements with mass or volume meters at the plant site should</del> To be cross-checked with sales receipts and an annual energy balance that is based on purchased quantities and stock changes
Any comment:	This variable is monitored directly in the site. It is the total amount of production. For instance, the sales receipts that contain the quantity sold may be used for monitoring

Data / Parameter table 3-4.

Data / Parameter:	$m_{pnr}, Q_{CO_2, y}$
Data unit:	t
Description:	Total amount of non-renewable biogenic residual CO <sub>2</sub> used in the process per tonne of inorganic compound produced in year y
Source of data:	Local measurements through field instruments On-site measurements
Measurement procedures (if any):	
Monitoring frequency:	Monthly
QA/QC procedures:	To be cross-checked with purchase/sales receipts
Any comment:	The amount of non-renewable CO <sub>2</sub> eventually used in the project needs to be monitored directly in the project site. The means of monitoring depends on each specific project. For instance, if CO <sub>2</sub> is purchased from external suppliers, then, this variable can be monitored from the amount of CO <sub>2</sub> purchased. The purchase receipts may be used for this purpose. Based on the carbon content (i.e. the number of carbon atoms in the compound molecule) and molecular weight of the inorganic compound produced in year y

Data / Parameter table 4.

Data / Parameter:	$m_{pr}$
Data unit:	t
Description:	Total amount of renewable biogenic residual CO <sub>2</sub> used in the process
Source of data:	Local measurements through field instruments
Measurement procedures (if any):	
Monitoring frequency:	Monthly
QA/QC procedures:	
Any comment:	This variable is calculated from $m$ and $m_{pnr}$ . The calculation depends on the chemical produced and the stoichiometric equation that represents its production. With the stoichiometric equation, and the monitored variables $m$ and $m_{pnr}$ , the calculation is performed as a conventional stoichiometric calculation

## 6.6. Baseline emission parameters

59. The following table illustrates the data to be collected or used in order to estimate emissions from the baseline activity.

Data / Parameter table 5.

Data / Parameter:	$m_{bnr}$
Data unit:	t
Description:	Total amount of non-renewable CO <sub>2</sub> used in the process before the start of the project activity
Source of data:	Project site records

Measurement procedures (if any):	
Monitoring frequency:	Monthly, over three years before the start of the project activity
QA/QC procedures:	
Any comment:	The amount of non-renewable CO <sub>2</sub> eventually used in the baseline needs to be monitored directly in the site before the start of the project activity. The means of monitoring depends on each specific project. In case that no renewable CO <sub>2</sub> has been used before the start of the project activity, $m_{br}$ has not to be monitored

**Data / Parameter table 6.**

Data / Parameter:	$m_{br}$
Data unit:	t
Description:	Total amount of renewable CO <sub>2</sub> used in the process before the start of the project activity
Source of data:	Project site records
Measurement procedures (if any):	
Monitoring frequency:	Monthly, over three years before the start of the project activity
QA/QC procedures:	
Any comment:	This variable is calculated from $m$ and $m_{br}$ . The calculation depends on the chemical produced and the stoichiometric equation that represents its production. With the stoichiometric equation, and the monitored variables $m$ and $m_{br}$ , the calculation is performed as a conventional stoichiometric calculation. In case that no renewable CO <sub>2</sub> has been used before the start of the project activity, $m_{br}$ is 0

**Data / Parameter table 7. 5.**

Data / Parameter:	<b>Product</b>
Data unit:	Description of chemical substance
Description:	Type of inorganic compound produced
Source of data:	Plant operator On-site records
Measurement procedures (if any):	
Monitoring frequency:	Annually after start of project activity
QA/QC procedures:	Type of inorganic compound well known to producers
Any comment:	The type of inorganic compound produced is monitored to assure ensure that the product does not change and that the methodology remains applicable during the crediting period

60. Monthly, project participants should provide the stoichiometric balance of the process or database of measured quantities. Field instruments must be regularly checked, calibrated, and maintained. The amounts monitored should be cross-checked with the legal purchase receipts of these products.

**6.7. Leakage**

61. The main potential source of leakage for this project activity lies in an increase in emissions due to diversion of CO<sub>2</sub> from other users to the project as a result of the project activity. This condition is checked in the validation of the project.

- - - - -

**Document information**

<i>Version</i>	<i>Date</i>	<i>Description</i>
03.0	9 September 2021	EB 111, Annex 6 Revision to: <ul style="list-style-type: none"> <li>clarify the source of baseline CO<sub>2</sub>;</li> <li>expand the eligibility of sources CO<sub>2</sub> in the baseline; and</li> <li>address provisions that could potentially lead to claiming emission reductions from CO<sub>2</sub> sequestration.</li> </ul>
02.0	29 September 2006	EB 26, Annex 10 Revision to clarify that the approved methodology is applicable to project activities where the generation of CO <sub>2</sub> from fossil or mineral sources in the baseline is only for the purpose of producing CO <sub>2</sub> used for the production of inorganic compounds and there is no energy by-product of CO <sub>2</sub> production from fossil source and its consumption in the baseline.
01.0	25 November 2005	EB 22, Annex 12 Initial adoption.
Decision Class: Regulatory Document Type: Standard Business Function: Methodology Keywords: biogas recovery, biomass, fuel switching		